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[Scale of cooperative \$\gamma\$ -relaxation of bisphenol A ...](#) - Xiao - Cited by 8

Detecting the endpoint in interfacial aromatic polycarbonate ...

The method of claim 1 wherein the **interfacial** polycarbonate **catalyst** is a ... The organic **solvent** can be any organic liquid which solubilizes **phosgene** and ...
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Process for producing polycarbonates from nitrogen diol and ...

The process of **interfacial** polycondensation using **phosgene** can produce high ... The **aromatic** diol haloformate is also in an organic **solvent** solution. ...
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In an **interfacial** polymerization, an. **aromatic bischloroformate** will be hydrolyzed in the ... the freezing of the aqueous **phase**). A. complex **catalyst** ...
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Thermoplastic resin compositions - US Patent 5455302

Component (B) is a **copolymer** which contains (a) an **aromatic** vinyl monomer ... After washing the organic **solvent phase** three times with 2% hydrochloric acid, ...
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Merrill described the **interfacial** copolymerization of poly(bisphenol-A ... As determined by GPC using THF as the **solvent**, the **copolymer** has a weight average ...
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The **pH** was decreased to about 9 with addition of. **phosgene**, and the aqueous **phase** was removed from. the reactor with suction. The **copolymer** solution ...
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[PDF] Silicone polycarbonates from new aryl-terminated siloxane precursors

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siloxane silicon polycarbonate **copolymer** polycarbonate **interfacial** ... solution at **pH** 9-11 using triethylamine as **phase**. transfer **catalyst**. **Phosgene** was ...
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amine or a **phase** transfer **catalyst**. **Phosgene** is then added to complete the ... M. Nishiguchi and T. Tokuda, **Aromatic** polycarbonate **copolymer**, a process for ...
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Siloxane-containing polymers - Patent Review 4499149

The **phase** transfer **catalyst** is believed to proceed as follows. ... a poly-basic carboxylic acid, anhydride or acyl halide, **phosgene**, or a **bischloroformate**. ...
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through an **interface** to make analysis, and calculating the data ... an aliphatic or **aromatic bischloroformate** in a methylene chloride **solvent** using pyridine ...
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chloride as a **catalyst** ... and the **copolymer**), but the thermal stability increases ... with the amide **solvents**, the dichlorides of **aromatic acids** ...www.turpion.org/php/full/infoFT.phtml?journal_id=rc&paper_id=1962&year_id=1971&volume_id=40&a... - [Similar pages](#)[\[PDF\] The Presidential Green Chemistry Challenge Awards Program Summary ...](#)

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within the **pH** range of 6-8. An **interface** layer is applied between the active ... **copolymer** is produced by polymerizing the acrylic monomer in **solvent** using ...www.ticinotransfer.ch/Documenti/pdf/RimanereInformati_Download/RIBEM_NEWS_2006-7.pdf - [Similar pages](#)[Polycarbonates suitable for use in optical articles - Patent ...](#)If the **interfacial** process is used, the addition of various **phase** transfer ... This process does not require the use of **phosgene** or a **solvent** and minimizes ...www.wikipatents.com/6001953.html - 180k - [Cached](#) - [Similar pages](#)[\[PDF\] Aliphatic Polyesters: Synthesis, Properties and Applications](#)

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tained (Scheme 1, a). a,w-**Bischloroformate** was then prepared by treating this. oligomer with **phosgene**. Poly(ester-carbonate) (Mn=30000 and Mw=48000) ...www.springerlink.com/index/B18U75BR2YGLTVUV.pdf - [Similar pages](#)[\[PDF\] Synthesis and Characterization of Cyclic Polymers and ...](#)

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the synthesis of **aromatic** polyethers. [49,50] . **Interfacial** condensation may be used ...The conditions (e.g. **solvent** quality, temperature, solid **phase**) are ...

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The yield of the **aromatic bischloroformate** was somewhat low and was ... filtered off and the **solvent** and excess **phosgene** stripped under. reduced pressure. ...

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Although addition of the **bischloroformate** neat (i.e., without **solvents**) is ... Preparation and Polymerization of a 4 mole % Cyclic Oligomers **Copolymer** of ...
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Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L1	963	bischloroformate and (diol or dihydroxy) and phosgene	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/12/18 09:34
L2	494	bischloroformate and (diol or dihydroxy) and phosgene and ph	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/12/18 09:34
L3	464	bischloroformate and aromatic and (diol or dihydroxy) and phosgene and ph	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/12/18 09:34
L4	192	bischloroformate and aromatic near4 (diol or dihydroxy) and phosgene and ph	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWENT; IBM_TDB	OR	ON	2006/12/18 09:34

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L5	185	bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ph	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:35
L6	182	bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ph and (polycarbonate or polysiloxane)	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:35
L11	144	bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and (acidity or alkalinity or ph) and (polycarbonate or polysiloxane) and copolymer	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:52
L12	144	bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and (acidity or alkalinity or ph) and (polycarbonate or polysiloxane) and copolymer	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:36
L13	141	bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and (polycarbonate or polysiloxane) and copolymer	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:52

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L14	20	aromatic near4 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and (polycarbonate or polysiloxane) and copolymer	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:52
L15	2	aromatic near4 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and (polycarbonate near3 polysiloxane) and copolymer	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:53
L16	20	aromatic near4 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and copolymer	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:54
L17	17	aromatic near4 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and copolymer and catalyst	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:54
L18	30	aromatic near10 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and copolymer and catalyst	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:55

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L19	30	aromatic near10 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and copolymer and catalyst and solvent	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:55
L20	29	aromatic near10 bischloroformate and aromatic near4 (\$8diol or dihydroxy\$8) and phosgene and ((acidity or alkalinity) near4 value or ph) and copolymer and catalyst and solvent and (phase or interfacial)	US-PGPU B; USPAT; USOCR; EPO; JPO; DERWEN T; IBM_TDB	OR	ON	2006/12/18 09:55